



Solution-phase combinatorial synthesis of isoxazolines and isoxazoles using [2+3] cycloaddition reaction of nitrile oxides

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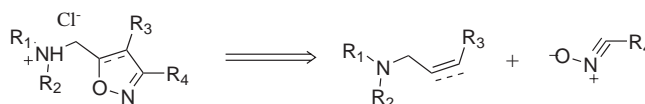
Abstract—An efficient way to construct a library of isoxazoles and isoxazolines was developed by solution-phase 1,3-dipolar cycloaddition reaction of nitrile oxides with olefins and alkynes followed by precipitation of the products as HCl salts. © 2001 Elsevier Science Ltd. All rights reserved.

Since combinatorial chemistry introduced a new paradigm in drug discovery,¹ combinatorial technology was expected to take a major role in post-genome projects² as well as in research for drug discovery,³ new materials⁴ and catalyst⁵ development. Solid-phase synthesis has been the major tool for combinatorial chemistry with apparent advantages over solution-phase chemistry.⁶ However, solution-phase combinatorial chemistry is emerging as a viable alternative to solid-phase synthesis due to the homogeneous nature and generality of solution phase chemistry.⁷

Herein, we wish to report an efficient solution-phase combinatorial strategy for the construction of a library of isoxazol(in)es. Since isoxazol(in)es were utilized as scaffolds for peptidomimetics⁸ and core structures in medicinal chemistry,⁹ several combinatorial synthesis of isoxazol(in)e libraries were reported. In most cases, however, most substitution possibilities of the isoxazol(in)e rings could not be realized since isoxazol(in)es were incorporated as building blocks instead of being constructed in combinatorial chemistry. To fully utilize the diversity feature of isoxazol(in)e ring, [2+3] cycloaddition reaction of nitrile oxides is most suitable. Unfortunately, solid-phase synthesis of isoxazol(in)es via nitrile oxides has not been successful due to low yields and low purities of the products.^{7a} We envisioned that the solution-phase 1,3-dipolar cycloaddition^{7c}

would be a good alternative to the solid-phase synthesis if the solid-solution biphasic nature of products and reagents of solid-phase synthesis could be implicated into the solution-phase synthesis for ready isolation of products from the reaction mixture. We added amine functionality to the isoxazol(in)e scaffold since amines could serve two purposes. One is to provide a facile means of purification and the other is to serve as a pharmacophoric group required for certain G-Protein Coupled Receptors (GPCR).¹⁰ Therefore every compound in the libraries contains an amine functionality in addition to the isoxazol(in)e ring. This feature allows four sites of attachment for pharmacophoric groups. The library was constructed through 1,3-dipolar cycloaddition reactions of nitrile oxides with dipolarophiles¹¹ (Scheme 1). The amine functionality in the scaffold is the key to solution-phase combinatorial chemistry as it provides an expeditious way of purification and isolation of the products from the reaction through precipitation of the products as HCl salts.

The library was constructed from cyclic amines attached to dipolarophiles and oximes (precursors of nitrile oxides) as listed in Fig. 1. The dipolarophiles



Scheme 1.

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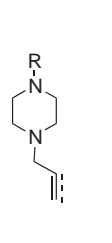
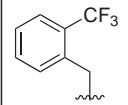
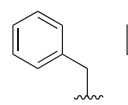
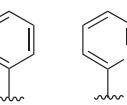
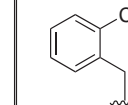
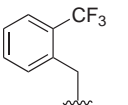
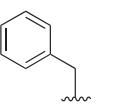
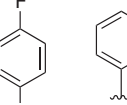
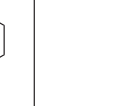
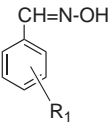
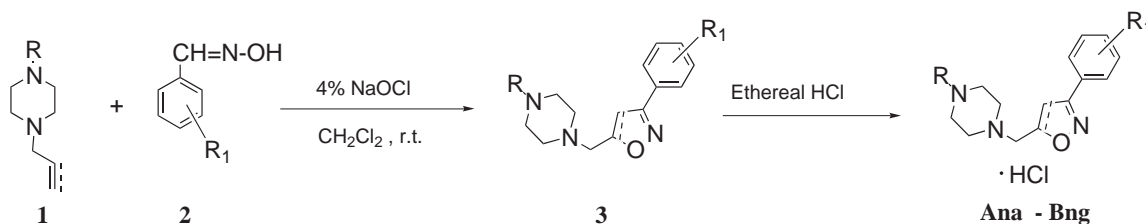
	R	Propargylic				Allylic			
		 A1	 A2	 A3	 A4	 B1	 B2	 B3	 B4
	R ₁	<i>p</i> -F	<i>p</i> -Cl	H	<i>p</i> -Me	<i>m</i> -OMe	<i>m</i> -Me	<i>o</i> -Cl, <i>o</i> '-F	
		a	b	c	d	e	f	g	

Figure 1.



Scheme 2.

were prepared from commercially available arylpiperazines with propargyl bromide or allyl bromide¹² and the oximes were prepared from aldehydes with hydroxylamine in presence of sodium hydrogen carbonate in EtOH/H₂O at 60°C.

With two scaffolds in hand, the library was constructed in a single operation via successive cycloaddition reaction followed by salt formation (Scheme 2).

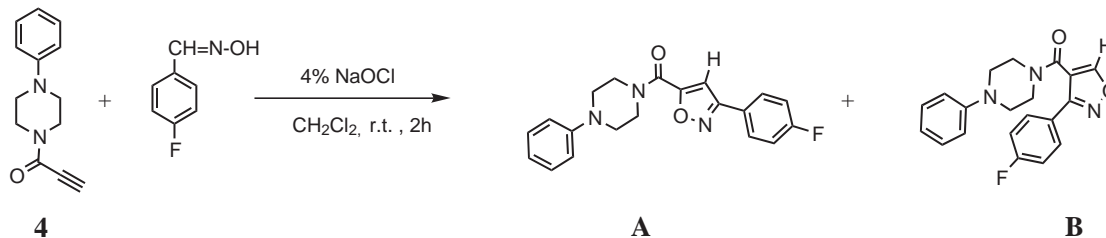
A solution of oxime(2, 5 equiv.) in dichloromethane was added to a solution of propargylic (or allylic) amine(1), sodium hypochlorite (4% aqueous solution (Aldrich), 5 equiv.) and Et₃N (1 equiv.). After the solvent was replaced to diethyl ether, the reaction mixture was treated with 1 M solution of HCl in diethyl ether. White solid was obtained, and was washed with diethyl ether several times. All the compounds were obtained in good yield (65% to 100%) with purities

Table 1. Isolated yields and purities^a of isoxazolines (A1–E5)

	a	b	c	d	e	f	g
A1	A1a 78.9(77.2) ²	A1b 80.7(78.8)	A1c 90.0(100)	A1d 90.0(100)	A1e 89.0(100)		
A2	A2a 46.1(100)	A2b 79.0(97.6)	A2c 93.9(72.7)	A2d 93.9(72.7)	A2e 52.5(84.5)		
A3	A3a 90.0(65.1)	A3b 40.2(100)	A3c 90.0(100)	A3d 90.0(100)	A3e 83.3(79.1)		
A4	A4a 54.8(69.2)	A4b 93.3(79.7)	A4c 90.0(70.5)	A4d 90.0(70.5)	A4e 92.2(75.1)		
B1	B1a 78.3(85.1) ^b	B1b 37.1(100)	B1c 87.3(85.5)			B1f 39.4(71.2)	B1g 72.9(79.8)
B2	B2a 37.8(72.2)	B2b 45.3(93.6)	B2c 48.7(84.6)			B2f 24.0(73.5)	B2g 78.3(89.2)
B3	B3a 76.3(88.6)	B3b 46.0(94.3)	B3c 75.8(81.6)			B3f 56.2(87.3)	B3g 92.3(94.1)
B4	B4a 39.8(86.7)	B4b 12.9(77.5)	B4c 70.9(82.0)			B4f 23.5(84.4)	B4g 77.4(100)

^a Determined by GC–MS or HPLC.

^b Yield (purities).



Scheme 3.

ranging from 70% to 100%. The result was quite compatible with general solid-phase synthesis as excess amount of nitrile oxides ensured complete conversion of **1** to **3** and did not pose severe contamination of the products. The purity and the identity of the products were confirmed by ¹H NMR and GC–MS or HPLC. All compounds were obtained as single isomeric products, though cycloaddition reaction could produce two regioisomeric products. Thus, a library of 20 isoxazoles and 20 isoxazolines was constructed (Table 1).

The main impurities were unreacted oxime derivatives and the impurities could be eliminated from the products by additional liquid–liquid extraction before the salt formation, though current protocol gave a satisfactory result.

An interesting result was observed when a conjugated alkyne, **4**¹³ was subjected to the reaction with oxime **a** (Scheme 3). Though we expected a single isomeric product formation based on the literature example,¹⁴ a mixture of regioisomeric products (**A**, **B** with 1.57:1 ratio)^{15,16} was obtained in 64.9% yield. This unexpected result is quite noteworthy since this reaction could easily double the size of a library of isoxazol(in)es with an advantage of producing spatially well differentiated regioisomeric products that could show distinctive biological activities.

In summary, 1,3-dipolar cycloaddition reaction of nitrile oxides to alkenes or alkynes was developed into a solution-phase combinatorial chemistry for the construction of a library of 20 isoxazoles and 20 isoxazolines with good yield and purity. This methodology can readily be expanded to a bigger library construction and will serve a complimentary method to the solid-phase combinatorial synthesis. The isoxazol(in)es with biogenic amines could serve as a part of 'privileged structures'¹⁷ for various receptors. Currently we are exploring to construct the mixture library of isomeric cycloaddition products and the application to lead a new search for GPCRs.

Acknowledgements

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References

1. *Combinatorial Chemistry*; Abelson, J. M., Ed. Meth. In Enzy. Academic Press: San Diego, 1996; Vol. 267.
2. Drews, J. *Science* **2000**, *287*, 1960–1964.
3. *Combinatorial Chemistry and Molecular Diversity in Drug Discovery*; Gordon, E. M.; Kerwin, Jr. J. F. Eds.; Wiley-Liss: New York, 1998.
4. Jandeleit, B.; Schaefer, D. J.; Powers, T. S.; Turner, H. W.; Weinberg, W. H. *Angew. Chem., Int. Ed.* **1999**, *38*, 2495–2532.
5. Poerter, J. R.; Wirschun, W. G.; Kuntz, K. W.; Snapper, M. L.; Hoveyda, A. H. *J. Am. Chem. Soc.* **2000**, *122*, 2657–2658.
6. Bunin, B. *The Combinatorial Index*; Academic Press: San Diego, 1998.
7. (a) *Combinatorial Chemistry: Synthesis, Analysis, Screening*; Jung, G., Ed., Wiley-VCH: Weinheim, 1999; (b) Baldino, C. M. *J. Comb. Chem.* **2000**, *2*, 89–103; (c) Studer, A.; Curran, D. P. *Tetrahedron* **1997**, *53*, 6681–6696.
8. Chung, Y. J.; Ryu, E. J.; Keum, G.; Kim, B. H. *Bioorg. Med. Chem.* **1996**, *4*, 209–225.
9. Confalone, P. N.; Jin, F.; Mousa, S. A. *Bioorg. Med. Chem. Lett.* **1999**, *9*, 55–58.
10. Watson, S.; Arkinstall, S. *The G-Protein Linked Receptor*; Academic Press: San Diego, 1994.
11. Padwa, A. *1,3-Dipolar Cycloaddition Chemistry*; John Wiley & Sons: New York, 1984; Vol. 1, p. 2.
12. Propargylic amines and allylic amines were prepared as follows. Propargyl (allyl) bromide (0.50 mol) was added over 20 min to a mixture of an amine (1.10 mol) in diethyl ether (500 ml). The reaction was stirred at room temperature for 18 hours. The resulting suspension was cooled to 0°C. The product was filtered and washed well with dry Et₂O and dried in vacuo.
13. 1-Phenylpiperazine (1 equiv.) and Et₃N (2 equiv.) was added to a stirred solution of propiolic acid (1.2 equiv.) and PCl₅ (1.2 equiv.) in CH₂Cl₂ at 40°C. After an hour of stirring, reaction mixture was washed with brine and water. Concentration followed by flash chromatography produced **4** in 67% yield.
14. Meyer, A. G.; Easton, C. J.; Lincoln, S. F.; Simpson, G. W. *J. Org. Chem.* **1998**, *63*, 9069–9075.
15. **Compound A**: ¹H NMR (300 MHz, CDCl₃): δ 7.82 (2H, dd, *J*=7 Hz, *J*=12 Hz), 7.33 (2H, t, *J*=9 Hz), 7.19 (2H, t, *J*=12 Hz), 7.12 (1H, s), 7.00 (3H, m), 3.99 (4H, br s), 3.32 (4H, br s). **Compound B**: ¹H NMR (300 MHz, CDCl₃): δ 8.65 (1H, s), 7.77 (2H, dd, *J*=6 Hz, *J*=12

Hz), 7.29 (2H, t, $J=9$ Hz), 7.19 (2H, t, $J=12$ Hz), 6.90 (3H, m), 3.94 (2H, br s), 3.39 (2H, br s), 3.22 (2H, br s), 2.80 (2H, br s). HRMS EI⁺, Anal. calcd for C₂₀H₁₈FN₃O₂: 351. 1383. Found: 351.1383.

16. Structural assignment of **A** and **B** are made based on the ¹H NMR of isoxazole protons and based on the assignment made in Ref. 14.
17. Patchett, A. A.; Nargund, R. P. *Annu. Rep. Med. Chem.* **2000**, 35, 289–298.